# SELENA AND THIADIAZOLE FUSED POLYCYCLICPOLYTHIA COMPOUNDS - PART-III

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Abstract: The selena and thiadiazole rings have been fused on 1/2-thiachroman-4-ones <u>1</u> & <u>2</u>, 4,9/4, 10-dithiahexahydrophenanthren-1-ones <u>9</u> & <u>10</u>, D-homo-6, 11/7, 11-dithia-1,3,5(10), 8,13(14)-gonapentaen-18a-ones <u>17</u> & <u>18</u> through their semicarbazones by reaction with selenium dioxide and thionyl chloride, respectively. The new compounds are characterized by thier 1R and <sup>1</sup>H NMR data.

## Introduction

In the chase for new and novel heterocycles capable of performing a variety of functions, steroid analogs have been identified as unique and interesting compounds. Polycyclicpolythia compounds which are analogous to thiasteroids exhibit a variety of chemotherapeutic properties (1-3). In continuation of our interest in polycyclicpolythia compounds (4,5), it was thought to develop selena/thiadiazole rings on the latter so that the resulting compounds would not only consists of sulfur but also nitrogen and/or selenium in the fused ring systems. Indeed, reports about selenium containing heterocycles which are biologically potent are relatively less (6-8). Therefore, the present study aims at the development of hitherto unknown selena and thiadiazole fused polycyclicpolythia compounds.

## **Results & Discussion**

The  $\alpha$ -ketomethylene group present in thiachroman-4-one <u>1</u> and 2-thiachroman-4-one <u>2</u> has been exploited to develop selena and thiadiazole rings. Treatment of <u>1</u> and <u>2</u> with semicarbazide hydrochloride gave the corresponding semicarbazones, <u>3</u> and <u>4</u>. The latter on reaction with selenium dioxide in acetic acid undergoes oxidative cyclization (6,7) to 1-thiadihydronaphthalen [3,4-d]1',2',3'-selenadiazole <u>5</u> and 2-thiadihydronaphthalen [3,4d]-1',2',3'-selenadiazole <u>6</u>. When <u>3</u> and <u>4</u> subjected to Hurd-Mori reaction process with thionyl chloride in dichloromethane (9) furnished 1-thiadihydronaphthalen [3,4-d]1',2',3'-

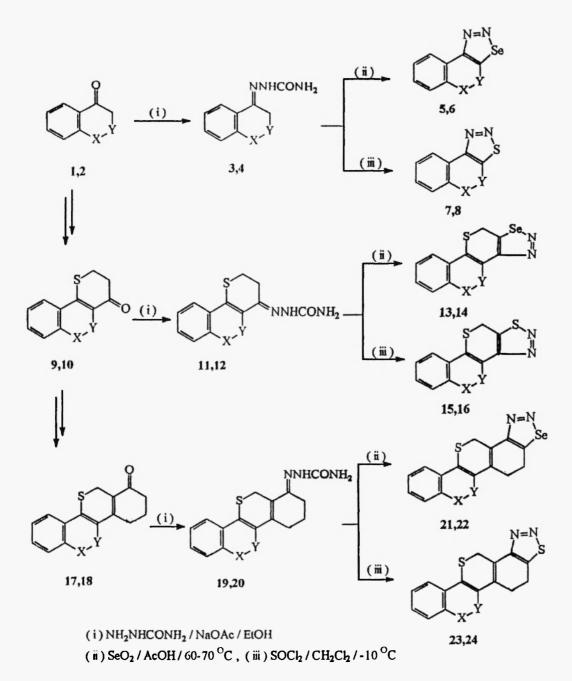
thiadiazole 7 (10) and 2-thiadihydronaphthalen [3,4-d]1,2,3-thiadiazole 8. On the other hand, condensation of 1 and 2 with  $\beta$ -mercaptopropanoic acid followed by cyclodehydration with phosphorus pentoxide furnished 4,9-dithiahexahydrophenanthren-1-one  $\underline{9}$ and 4,10-dithiahexahydrophenanthren-1-one 10, respectively (4). The selena and thiadiazole rings have been developed on the latter by oxidative cyclization with selenium dioxide and Hurd-Mori reaction with thionyl chloride via their semicarbazones 11 & 12. Thus 4.9dithiatetrahydrophenanthren [1,2-d] 1',2',3'-selena/ thiadiazole 13 & 15 and 4,10dithiatetrahydrophenanthren [1,2-d] 1',2',3'-selena/thiadiazole 14 & 16 have been prepared. Besides this, D-homo-6,11-dithia-1,3,5(10),8,13(14)-gonapentaen-18a-one 17 and D-homo-7,11-dithia-1,3,5(10),8,13(14)-gonapentaen-18a-one 18 have been obtained by the reaction of <u>9</u> and <u>10</u> with  $\gamma$ -chlorobutyric acid followed by cyclodehydration with phosphorus pentoxide (4). The selena and thiadiazole rings have been developed on 17 and 18 by adopting the above methodology so as to get D-homo-6,11-dithia-1,3,5(10),8,13(14)gonapentaen [18,18a-d]1', 2', 3'-selena/thiadiazole (21 & 23) and D-homo-7,11-dithia-1,3,5(10),8,13(14)-gonapentaen[18,18a-d]1',2',3'-selena thiadiazole (22 & 24), a new class of polycyclicpolythia selena/thiadiazoles (Scheme). All the compounds were obtained in relatively good yields (62 -78%) and were chromatographically pure.

### Experimental

Melting points were determined on a Mel-Temp apparatus which are uncorrected. The IR spectra in KBr pellets were recorded on a Perkin-Elmer 993 infrared spectrometer ( $\nu$  in cm<sup>-1</sup>), <sup>1</sup>H NMR spectra were run in CDCl<sub>3</sub> on 90 or 200 MHz on Perkin-Elmer instrument with TMS as an internal standard (chemical shifts in  $\delta$ , ppm).

Preparation of semicarbazones of thiachroman-4-one  $\underline{3}$ , 2-thiachroman-4-one  $\underline{4}$ , 4,9dithiahexahydrophenanthren-1-one  $\underline{11}$ , 4,10-dithiahexahydrophenanthren-1-one  $\underline{12}$ , D-homo 6,11-dithia-1,3,5(10),8,13(14)-gonapentaen-17a-one  $\underline{19}$  and D-homo-7,11dithia-1,3,5(10), 8,13(14)-gonapentaen-18a-one  $\underline{20}$ 

A mixture of  $\frac{1/2}{9}/\frac{10}{17}/18$  (0.005 mole), semicarbazide hydrochloride (0.001 mole) sodium acetate (0.002 mole) in ethanol (40 ml) was refluxed for 2-3h on a steam-



SCHEME

bath and cooled. The separated solid was filtered, washed with water and recrystallized from ethanol to get pure 3/4/11/12/19/20. 3: Yield 84%, m.p. 156-57°C; 4: Yield 81%, m.p. 165-66°C; 11: Yield 80%, m.p. 205-08°C; 12: Yield 76%, m.p.197-99°C; 19: Yield 72%, m.p. >300°C; 20: Yield 75%, m.p. >300°C.

Preparation of 1-thiadihydronaphthalen [3,4-d] 1',2',3'-selenadiazole 5, 2-thiadihydronaphthalen [3,4-d] 1',2',3'-selenadiazole 6, 4,9-dithiatetrahydrophenanthren-[1,2-d]1',2',3'-selenadiazole 13, 4-10-dithiatetrahydrophenanthren [1,2-d] 1',2',3'selenadiazole 14, D-homo-6,11-dithia-1,3,5(10),8,13(14)-gonapentaen[18,18a-d] 1',2',3'-selenadiazole 21, D-homo-7,11-dithia-1,3,5(10),8,13(14)-gonapentaen[18,18ad] 1',2',3'-selenadiazole 22.

The semicarbazone 3/4/11/12/19/20 (0.0005) mole was treated with selenium dioxide powder (0.0005) mole in glacial acetic acid (20 ml) and the mixture was gently heated (60-70°) with stirring unitl the evolution of gas ceased. Then the reaction mixture was cooled, filtered and poured onto crushed ice, to get a solid product. It was filtered, dried and recrystallized from pet. ether (60-80°C) to afford a pure compound. 5: Yield 66%, m.p. 98-100°C (Found C, 42.53; H, 2.51, C<sub>9</sub>H<sub>6</sub>N<sub>2</sub>SSe requires C, 42.70, H, 2.39%); IR : 1450 (N=N) cm<sup>-1</sup>, <sup>1</sup>H NMR: δ 4.28 (s, 2H, CH<sub>2</sub>), 6.85-7.02 (m, 4H, Ar-H). <u>6</u>: Yield 62%, m.p. 95-97° C (Found C, 42.57; H, 2.28, C<sub>9</sub> H<sub>6</sub>N<sub>2</sub>SSe requires C,42.70; H, 2.39%); IR: 1445 (N=N) cm<sup>-1</sup>; <sup>1</sup>H NMR: δ 4.25 (s, 2H, CH<sub>2</sub>), 6.91-7.05 (m, 4H, Ar-H). <u>13</u>: Yield 69%, m.p. 123-25°C (Found C, 44.80; H, 2.36, C<sub>12</sub>H<sub>8</sub>N<sub>2</sub>S<sub>2</sub>Se requires C, 44.58; H 2.49%); IR: 1448 (N=N) cm<sup>-1</sup>, <sup>1</sup>H NMR: δ 4.18 (s, 2H, S-CH<sub>2</sub> - C-Se), 4.28 (s, 2H, CH<sub>2</sub>), 7.02-7.14 (m, 2H, Ar-H). 14: Yield 65% m.p. 131-33°C (Found C, 44.77; H, 2.65 C<sub>12</sub>H<sub>8</sub>N<sub>2</sub>S<sub>2</sub>Se requires C, 44.58: H, 2. 49%); IR: 1452 (N=N) cm<sup>-1</sup>; <sup>1</sup>H NMR: δ 4.21 (s, 2H, S-CH<sub>2</sub>-C-Se), 4.29 (s, 2H, CH<sub>2</sub>), 7.05-7.15 (m, 4H, Ar-H). 21: Yield 60%, m.p. 143-45°C (Found C, 50.94; H, 3.30, C<sub>16</sub>H<sub>12</sub>N<sub>2</sub>S<sub>2</sub>Se requires C, 51.20, H, 3. 22%), IR: 1438 (N=N) cm<sup>-1</sup>, <sup>1</sup>H NMR: δ 2.98-3.12 (m, 4H, 2CH<sub>2</sub>), 4.24 (s, 2H, S-CH<sub>2</sub>-C=C), 4.30 (s, 2H, Ar-S-CH<sub>2</sub>-C=C), 7.00-7.12 (m, 4H, Ar-H). 22: Yield 64%, m.p. 137-39°C (Found C,

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51.42; H, 3.39,  $C_{16}H_{12}N_2S_2Se$  requires C, 51.20; H, 3. 22%); IR: 1440 (N=N) cm<sup>-1</sup>; <sup>1</sup>H NMR:  $\delta$  2.85-3.02 (m, 4H, 2CH<sub>2</sub>), 4.25 (s, 2H, S-CH<sub>2</sub>-C=C), 4.34 (s, 2H, Ar-CH<sub>2</sub>-S-C=C), 6.95-7.03 (m, 4H, Ar-H).

Preparation of 1-thiadihydronaphthalen [3,4-d]1',2',3'-thiadiazole  $\underline{7}$ , 2-thiadihydronaphthalen [3,4-d]1',2',3'-thiadiazole  $\underline{8}$ , 4,9-dithiatetrahydrophenanthren [1,2-d]1',2',3'-thiadiazole  $\underline{15}$ , 4,10-dithiatetrahydrophenanthren [1,2-d]1',2',3'-thiadiazole  $\underline{16}$ , D-homo-6, 11-dithia-1,3,5(10), 8,13(14)-gonapentaen [18,18a-d]1',2',3'-thiadiazole  $\underline{23}$ , D-homo-7, 11-dithia-1,3,5(10), 8,13(14)-gonapentaen [18,18a-d]1',2',3'-thiadiazole  $\underline{24}$ .

The semicarbazone 3/4/11/12/19/20 (0.0005 mole) was added portion-wise to thionyl chloride (1.5 ml) at ice-bath temperature and later kept for 1h at room temperature. Then dichloromethane (10 ml) was added and decomposed with ice-cold sodium carbonate solution. The organic layer was separated, washed with water (10 ml, 4-5 times) and dried over anhydrous sodium sulfate. Evaporation of the solvent furnished a gummy product which was solidified by treatment with cyclohexane. It was purified by recrystallization from ethanol to get a pure compound. 7: Yield 68%, m.p. 71-73°C (lit m.p. 72-75°C). 8: Yield 70%, m.p. 106-07°C (Found C, 52. 26; H, 2.80; C<sub>9</sub> H<sub>6</sub>N<sub>2</sub>S<sub>2</sub> requires C, 52.40; H, 2.93%) IR: 1439 (N=N) cm<sup>-1</sup>, <sup>1</sup>H NMR; δ 4.28 (s, 2H, CH<sub>2</sub>) 7.02-7.15 (m, 4H, Ar-H). <u>15</u>: Yield 65%, m.p. 101- 02° C (Found C, 51. 89; H, 2.79, C<sub>12</sub>H<sub>8</sub>N<sub>2</sub>S<sub>3</sub> requires C, 52.14; H, 2.91%); **IR**: 1450 (N=N) cm<sup>-1</sup>; <sup>1</sup>H NMR: δ 4.24 (s, 2H, S-CH<sub>2</sub>-C-S), 4.27 (s, 2H, CH<sub>2</sub>), 6.98-7.12 (m, 4H, Ar-H). 16: Yield 68%, m.p.114-15°C (Found C, 51.93; H, 2.82,  $C_{12}H_8N_2S_3$  requires C, 52.14; H, 2.91%); IR: 1452 (N=N) cm<sup>-1</sup>; <sup>1</sup>H NMR:  $\delta$  4.25 (s, 2H, S-CH<sub>2</sub>-C-S), 4.31 (s, 2H, CH<sub>2</sub>), 7.02-7.13 (m, 4H, Ar-H). 23: Yield 60% m.p. 112-13°C (Found, C, 58.70, H, 3.57; C<sub>16</sub>H<sub>12</sub>N<sub>2</sub>S<sub>3</sub> requires C, 58.51; H, 3.68%); IR:1440 (N=N) cm<sup>-</sup> <sup>1</sup>; <sup>1</sup>H NMR: δ 2.89-3.05 (m, 4H, 2CH<sub>2</sub>), 4.28 (s, 2H, S-CH<sub>2</sub>-C=C), 4.34 (s, 2H, Ar-S-CH2-C=C), 6.95-7.04 (m, 4H, Ar-H). 24: Yield 64%, m.p. 107-08°C (Found C, 58.40, H, 3.82;  $C_{16}H_{12}N_2S_3$  requires C, 58.51; H, 3.68%); IR: 1450 (N=N) cm<sup>-1</sup>; <sup>1</sup>H NMR:  $\delta$ 

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2.91-3.09 (m, 4H, 2CH<sub>2</sub>), 4.25 (s, 2H, S-CH<sub>2</sub>-C=C), 4.36 (s, 2H, Ar-CH<sub>2</sub>-S-C=C), 6.98-

7.08 (m, 4H, Ar-H).

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